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In situ EC-AFM observation with atomic resolution of Pb(1 0 0) and Pb(1 1 1) single crystals in sulfuric acid solution

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Abstract

Pb(1 0 0) and Pb(1 1 1) single crystals in 50 mM H₂SO₄ aqueous solution have been investigated by in situ electrochemical atomic force microscopy (EC-AFM) under potential control. We have succeeded in observating high resolution images of Pb(1 0 0) and Pb(1 1 1) single crystals, as well as high resolution and morphological images of PbSO₄ crystals formed on the Pb single crystals. We found that the top face of PbSO₄ crystals formed on Pb(1 0 0) is PbSO₄(0 0 1), while that on Pb(1 1 1) is PbSO₄(1 0 0). We also found that the atomic rows along the [1 0 0] and [0 1 0] direction of PbSO₄(0 0 1) surface were parallel with those along the [0 1 1] and [0 1 $\overline{1}$] direction of the Pb(1 0 0) substrate, while those along the [0 1 0] and [0 0 1] direction of the PbSO₄ (1 0 0) surface were parallel with those along the [0 1 $\overline{1}$] and [$\overline{1}$ 1 1] direction of the Pb(1 1 1) substrate.

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1. Introduction

The behavior of Pb electrodes in sulfuric acid solution is very important from both scientific and practical viewpoints because it closely relates to the electrochemical reaction on the negative electrode of lead–acid batteries. Electrochemical atomic force microscopy (EC-AFM), as well as electrochemical scanning tunneling microscopy (EC-STM), is one of the powerful tools, which can observe the morphology of the electrode surfaces in electrolyte. Remarkable knowledge about the electrochemical reactions on the negative and positive electrodes of lead–acid batteries has been obtained through our EC-AFM observation of the morphological changes on Pb and PbO₂ electrodes in sulfuric acid solution during oxidation and reduction [1–5].

On the other hand, EC-AFM and EC-STM can also observe atomic or molecular arrangements on single crystals of noble metals in electrolyte [6,7]. However, there are few atomic resolution EC-AFM or EC-STM observations of the electrodes of base metals, such as nickel [8,9], cobalt [10], iron [11], lead [12], and so on, because it is generally difficult to remove the surface oxide layer formed in air. In this paper, we have investigated the in situ EC-AFM observations of Pb(1 0 0) and Pb(1 1 1) single crystals in sulfuric acid solution under potential control.

2. Experimental

The samples used are Pb(1 0 0) and Pb(1 1 1) single crystals (MaTecK Co.). Immediately before the EC-AFM observation, the samples were polished chemically with a mixed solution of acetic acid and hydrogen peroxide and were washed finally with ethanol. The EC-AFM used is a Nanoscope IIIa (Digital Instruments). Fig. 1 shows the cross-section of EC-AFM cell. The potential of working electrode was controlled by a potentiostat and was referred to the reference electrode. The reference electrode used here was the Hg/Hg₂SO₄ electrode (0.65 V versus normal hydrogen electrode; NHE), to which all potential was referred in this paper. The space surrounded by the working electrode, the cantilever mount and the O-ring was filled with the electrolytes. The electrolytes used here were 50 mM H₂SO₄ aqueous solution, which were prepared from H₂SO₄ (Wako,

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Fig. 1. Schematic illustration of cross-section of EC-AFM electrochemical cell.







Fig. 3. EC-AFM morphological image $(5 \,\mu m \times 5 \,\mu m)$ on Pb(100) in 50 mM H₂SO₄ solution at $-1400 \,m$ V. The orientation of Pb(100) substrate is indicated by arrows.



Fig. 4. (a) EC-AFM high resolution image $(3 \text{ nm} \times 3 \text{ nm})$ on Pb $(1 \ 0 \ 0)$ in 50 mM H₂SO₄ solution at -1400 mV. The orientation of Pb $(1 \ 0 \ 0)$ substrate is indicated by arrows. (b) Schematic illustration of image (a).



Fig. 5. EC-AFM morphological image $(2.5 \,\mu\text{m} \times 2.5 \,\mu\text{m})$ on Pb(1 0 0) in 50 mM H₂SO₄ solution at 0 mV. The orientation of Pb(1 0 0) substrate is indicated by arrows.

Superior) and Milli-Q water. All of the experiments are performed at room temperature.

3. Results and discussion

3.1. Cyclic voltammogram

Fig. 2 shows cyclic voltammograms (CVs) for Pb(1 0 0) and Pb(1 1 1) in 50 mM H₂SO₄ solution with sweep rate of 1 mV s⁻¹. In the positive scan, there are oxidation peaks at -850 mV, which correspond to the reaction of Pb + SO₄²⁻ \rightarrow PbSO₄+2e⁻. Reversing the potential toward negative at



Fig. 7. Schematic illustration of orientation relationship between Pb atoms in Fig. 4 and SO₄ groups in Fig. 6. The orientation of Pb(1 0 0) substrate is indicated by arrows.

-650 mV, there are cathodic peaks at -1000 mV, which correspond to the reaction of $PbSO_4 + 2e^- \rightarrow Pb + SO_4^{2-}$. These CVs agree well with the one for the lead sheet samples [2].

3.2. EC-AFM observation on Pb(1 0 0)

Fig. 3 shows an EC-AFM morphological image (5 μ m × 5 μ m) on Pb(1 0 0) in 50 mM H₂SO₄ solution at -1400 mV. On the terrace in Fig. 3, a filtered high resolution EC-AFM image (3 nm × 3 nm) is obtained as shown in Fig. 4a. We can recognize the spots having four-fold symmetry with interatomic distance of 0.35 nm. It corresponds well to a bare and unreconstructed Pb(1 0 0) (1 × 1) structure as



Fig. 6. (a) EC-AFM high resolution image $(3 \text{ nm} \times 3 \text{ nm})$ on Pb $(1 \ 0 \ 0)$ in 50 mM H₂SO₄ solution at 0 mV. The orientation of Pb $(1 \ 0 \ 0)$ substrate is indicated by arrows. (b) Schematic illustration of image (a).



Fig. 8. Orientation relationship between Fig. 5 and SO_4 groups in Fig. 6. The orientation of Pb(1 0 0) substrate is indicated by arrows.

shown schematically in Fig. 4b. We have also succeeded in the observation of the growth of a PbSO₄ crystal on the Pb(1 0 0) electrode when the potential is stepped from -1400 to 0 mV, as shown in Fig. 5 (2.5 µm × 2.5 µm). On the top plane of the PbSO₄ crystal in Fig. 5, a filtered high resolution EC-AFM image (3 nm × 3 nm) is obtained as shown in Fig. 6a. We can recognize the spots, which have two-fold symmetry with interatomic distances of 0.41 and 0.52 nm. This superlattice corresponds to the protuberances by SO₄ groups of the unreconstructed PbSO₄(0 0 1) (1 × 1) structure (Fig. 6b). Fig. 7 shows a schematic illustration of orientation relationship between Pb atoms in Fig. 4 and SO₄ groups in Fig. 6. We found that the atomic rows along the [1 0 0] and [0 1 0] direction of $PbSO_4(0 0 1)$ surface were parallel with those along the [0 1 1] and [0 11] direction of Pb(1 0 0) substrate, respectively. Fig. 8 shows an orientation relationship between Fig. 5 and SO₄ groups in Fig. 6. We also found that the top plane of PbSO₄ crystal in Fig. 5 is surrounded by [0 1 1] and [0 11] steps.

3.3. EC-AFM observation on Pb(1 1 1)

Fig. 9a shows a filtered high resolution EC-AFM image $(2.5 \text{ nm} \times 2.5 \text{ nm})$. We can recognize the spots having three-fold symmetry with interatomic distance of 0.36 nm. It corresponds well to a bare and unreconstructed Pb(1 1 1)



Fig. 9. (a) EC-AFM high resolution image (2.5 nm \times 2.5 nm) on Pb(1 1 1) in 50 mM H₂SO₄ solution at -1400 mV. The orientation of Pb(1 1 1) substrate is indicated by arrows. (b) Schematic illustration of image (a).



Fig. 10. (a) EC-AFM high resolution image $(2.1 \text{ nm} \times 2.1 \text{ nm})$ on Pb $(1 \ 1 \ 1)$ in 50 mM H₂SO₄ solution at 0 mV. The orientation of Pb $(1 \ 1 \ 1)$ substrate is indicated by arrows. (b) Schematic illustration of image (a).

 (1×1) structure as shown schematically in Fig. 9b. We have also succeeded in the observation of the growth of a PbSO₄ crystal on the $Pb(1 \ 1 \ 1)$ electrode when the potential is stepped from -1400 to 0 mV, as shown in Fig. 12 $(2.5 \,\mu\text{m} \times 2.5 \,\mu\text{m})$. On the top plane of the PbSO₄ crystal in Fig. 12, a filtered high resolution EC-AFM image $(2.1 \text{ nm} \times 2.1 \text{ nm})$ is obtained as shown in Fig. 10a. We can recognize the spots, which have two-fold symmetry with interatomic distances of 0.53 and 0.68 nm. This superlattice corresponds to the protuberances by SO₄ groups (only at higher position) of the unreconstructed PbSO₄(1 0 0) (1 \times 1) structure (Fig. 10b). Fig. 11 shows a schematic illustration of the orientation relationship between Pb atoms in Fig. 9 and SO_4 groups (only at higher position) in Fig. 10. We found that the atomic rows along the [0 1 0] and [0 0 1] direction of PbSO₄ (100) surface were parallel with those along the $[0\ 1\overline{1}]$ and $[\overline{2}1\ 1]$ direction of Pb(1\ 1\ 1) substrate, respectively. Fig. 12 shows an orientation relationship between the



Fig. 11. Schematic illustration of orientation relationship between Pb atoms in Fig. 9 and SO_4 groups at higher position in Fig. 10. The orientation of Pb(1 1 1) substrate is indicated by arrows.



Fig. 12. EC-AFM morphological image $(2.5 \,\mu\text{m} \times 2.5 \,\mu\text{m})$ on Pb(1 1 1) in 50 mM H₂SO₄ solution at 0 mV and orientation relationship between this image and SO₄ groups at higher position Fig. 10. The orientation of Pb(1 1 1) substrate is indicated by arrows.

EC-AFM morphological image of a PbSO₄ crystal on the Pb(1 1 1) electrode and SO₄ groups (only at higher position) in Fig. 10. We also found that the top plane of the PbSO₄ crystal in Fig. 12 is surrounded by $[0 \ 1\bar{1}]$ and $[\bar{2}1 \ 1]$ steps.

4. Conclusion

Pb(1 0 0) and Pb(1 1 1) single crystals in 50 mM H_2SO_4 aqueous solution have been investigated by in situ EC-AFM under potential control. We have succeeded in observation on high resolution images of Pb(1 0 0) and Pb(1 1 1) single crystals, as well as high resolution and morphological images of PbSO₄ crystals formed on the Pb single crystals. The following conclusions were derived from Section 3.

- 1. We found that the top face of $PbSO_4$ crystals formed on $Pb(1\ 0\ 0)$ is $PbSO_4(0\ 0\ 1)$, while that on $Pb(1\ 1\ 1)$ is $PbSO_4(1\ 0\ 0)$.
- 2. We also found that the atomic rows along the $[1\ 0\ 0]$ and $[0\ 1\ 0]$ direction of PbSO₄ (0 0 1) surface were parallel with those along the $[0\ 1\ 1]$ and $[0\ 1\overline{1}]$ direction of Pb(1 0 0) substrate, while those along the $[0\ 1\ 0]$ and $[0\ 0\ 1]$ direction of PbSO₄ (1 0 0) surface were parallel with those along the $[0\ 1\overline{1}]$ and $[\overline{2}1\ 1]$ direction of PbSO₄ (1 0 1) surface were parallel with those along the $[0\ 1\overline{1}]$ and $[\overline{2}1\ 1]$ direction of Pb(1 1 1) substrate.

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